The Free Electron Gas Model

SFI5769 - Chemical Physics and Thermodynamics of Gases and Solids

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Cartoon image of Sommerfeld, Dirac and Fermi. Produced using ChatGPT.

Introduction

- The Model
 - Quantum Well
 - The Fermi Sphere and Fermi Energy
 - Density of States and the Fermi-Dirac Distribution
- **Consequences and Applications**
- Limitations and Failures

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Consequences and Applications

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A little bit of the history:

- Drude firstly proposed it, but in a classical and statistical physics view.
- Pauli principle and Fermi-Dirac statistics.
- Sommerfeld incorpored quantum effects.





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Understand properties of solids, mainly metals.

Simple ideia that explains a lot of fundamental

Crucial to the development of the band theory.

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Quantum Well



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Source: C. Kittel, Introduction to solid state physics, 8th ed. Hoboken, NY: Wiley (2005)

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Expandind to 3 dimensions...

Born-von Karman periodic boundary conditions (cube of L side inifinity lattice)



https://www.pixelsquid.com/png/cube-grid-patterned-black-3078631786566128952?image=G03

plane wave solution

where

 k_x

and the energy

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$$\psi_{ec{k}}(ec{r}) = rac{1}{\sqrt{V}} e^{iec{k}\cdotec{r}}$$

$$=rac{2\pi n_x}{L},\quad k_y=rac{2\pi n_y}{L},\quad k_z=rac{2\pi n_z}{L}$$

$$E_{ec{k}} = rac{m{\hbar}^2 k^2}{2m} = rac{m{\hbar}^2}{2m}ig(k_x^2 + k_y^2 + k_z^2ig)$$



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The Fermi Sphere and Fermi Energy



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Density of States and the Fermi-Dirac Distribution

The density of states, g(E), is defined as the <u>number</u> of available electronic states per unit volume, per unit <u>energy range.</u>

$$N(E) = rac{V}{3\pi^2} igg(rac{2mE}{\hbar^2}igg)^{3/2} \ = rac{V}{3\pi^2} rac{(2m)^{3/2}}{\hbar^3} E^{3/2}$$

$$egin{aligned} g(E) &= rac{1}{V} rac{dN(E)}{dE} = rac{d}{dE} iggl[rac{1}{3\pi^2} (2m)^{3/2} rac{1}{\hbar^3} E^{3/2} iggr] g(E) &= rac{1}{2\pi^2} iggl(rac{2m}{\hbar^2} iggr)^{3/2} \sqrt{E} \end{aligned}$$

density of states



Fermi-dirac distribution function:

chemical potential.

$$T=0$$
 , all er for $E < E_F$.

of the Fermi sphere.

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$$f(E)=rac{1}{e^{(E-\mu)/k_BT}+1}$$

where k_B is the Boltzmann constant and μ is the

nergy states completely filled f(E) = 1

T > 0, some electrons gain energy and smeares out

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 $pprox k_{ extsf{b}}T$



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Electrical Condutivity and Ohm's Law

Drude model: $m \frac{d \vec{v}_d}{d t} = -e \vec{E} - \frac{m \vec{v}_d}{\tau}$ solving $ec{v}_d = -rac{e au}{m}ec{E}$ Relating \vec{v}_d and \vec{j} we get the Ohm's Law: m

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 $C_{\mathrm{el}} =$



The number of "active thermal electrons" are the out of the Fermi sphere

$$\Delta U pprox N_{
m eff} \cdot (k_B T) pprox g(E_F) (k_B T)^2$$

using
$$C_{el}=dU/dT$$

$$rac{\pi^2}{3}g(E_F)k_B^2T=rac{\pi^2}{2}Nk_B\left(rac{T}{T_F}
ight)$$

dependence on T (!)

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Consequences and Applications

Thermal Conductivity and the Wiedemann-Franz Law

In metals, heat is primarily transported by the same free electrons that conduct electricity.

The thermal conductivity k can be estimated from kinetic theory.

The Wiedemann-Franz law is an empirical observation that the ratio of thermal to electrical conductivity for metals is directly proportional to temperature.

 $L = \frac{\kappa}{\sigma T}$

what is a great result for the model, because agrees with the experimental values.

Computing for metals, we got

 $L = 2.44 \times 10^{-8} W \Omega K^{-2}$

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Existence of insulators

- Accoding to the model → every element with valence electrons should be a metal.
- It cannot answer why Sodium is a metal while Diamond is an insulator.
- The concept of an energy band gap, which is the defining characteristic of semiconductors and insulators, is completely absent.

Hall Effect

For all metal the hall coeficient must be negative (all carriers are electrons)

some experiments shows that some metals has a positive Hall coeficient.

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$$R_H = -rac{1}{ne}$$

carriers \rightarrow holes!

Free Path Paradox

Model \rightarrow angstroms of distance of the ions, free path in this order.

Experimental \rightarrow tens to hundreds of angstroms (!)

The model fails to explain why electrons appear to ignore the vast majority of the ions in the lattice. This paradox is solve using the band theory.

The arrangement of atoms in a crystal is a direct result of the complex interplay between the electron wavefunctions and the discrete ion potentials.

This phenomenon the free electron model cannot capture.

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Cohesion and Crystal Structure

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Thanks for your attention!

Any questions?